

CUMAMBRIN A - A SESQUITERPENE LACTONE

FROM *Handelia trichophylla*

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We subjected dried inflorescences of the plant *Handelia trichophylla* Heimerl, family Compositae, growing in the Tashkent oblast, to exhaustive extraction with chloroform. The resin after the elimination of the solvent was triturated with hexane and ether; the resulting insoluble powder was chromatographed on neutral inactive alumina.

From a benzene eluate we isolated compound (I) with the composition $C_{17}H_{22}O_5$, mp 186-187°C (from ethanol), mol. wt. 306 (mass spectrometry), R_f 0.36 in the benzene-methanol (8:2) system on Silufol plates. The IR spectrum (tablets with KBr) has absorption bands at (cm^{-1}) 3520 (OH), 1755, 1625 (carbonyls of a γ -lactone and of an ester group), and 1670 (C=C).

On the basis of the features of the NMR, IR, and mass spectra of (I) and the products of its saponification, hydrogenation, and dehydrogenation, we have proposed for the lactone a structure coinciding with the structural formula of cumambrin A [1]. However, different melting points are given for this: 178°C [1] and 188-190°C with an acetyl derivative having mp 198-200°C [2].

A solution of compound (I) in acetyl chloride was kept in a sealed tube at 35-40°C for two days. This gave the acetyl derivative of the lactone (at a tertiary hydroxy) with the composition $C_{19}H_{24}O_6$, mp 197°C, IR spectrum, cm^{-1} : λ_{max}^{KBr} 1765, 1736, 1250 (γ -lactone and ester), 1660, 1630 (C=C).

Mixtures of compound (I) with samples of cumambrin A kindly given to us by J. Romo and T. A. Geissman melted with no depression of the temperature; the substances were also chromatographically identical.

Consequently, the lactone that we have isolated is identical with cumambrin A.

LITERATURE CITED

1. J. Romo, A. Romo de Vivar, and E. Dias, *Tetrahedron*, **24**, 5625 (1968).
2. M. A. Irwin and T. A. Geissman, *Phytochem.*, **305**, 8 (1969).

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